

## *Physical Properties Yahoo Answers*







### Physical Properties Yahoo Answers

Best Answer: They are all physical. Nothing is changing it's chemical form. Ice melts into water, but it is still H<sub>2</sub>O. Ditto with evaporating and freezing. Sand does not dissolve, it is just held up mechanically in the water and will settle out, but it is still just sand and water.

### Physical/Chemical Properties!? | Yahoo Answers

Can you help me with these questions? What physical properties would you use to separate these mixtures? 1) Iron fillings and salt 2) Salt and water Name two physical properties that could be used to distinguish between these substances. 3) Water and rubbing alcohol 4) Gold and aluminum 5) Helium gas and oxygen gas Thank you for your help ^\_^

### Physical Properties? | Yahoo Answers

Physical properties can be observed or measured without changing the composition of matter. Physical properties are used to observe and describe matter. Physical properties include: appearance, texture, color, odor, melting point, boiling point, density, solubility, polarity, and many others.

### what is physical property? | Yahoo Answers

Phosphorus trichloride is a liquid with a boiling point of 74C, it does not conduct electricity. Calcium chloride is a solid with a melting point of 772C; when molten it conducts electricity. Explain, in terms of bonding, why these compounds have such different properties. Phosphorus trichloride is a covalent bond and Calcium chloride is ionic.

### chemical bonding and physical properties? | Yahoo Answers

Yahoo Canada Answers ... Answers. Relevance. Rating Newest Oldest. Best Answer: Physical Properties Metric English Comments Density 1.54 g/cc 0.0556 lb/in<sup>3</sup> Chemical Properties Metric English Comments Atomic Number 20 20 Thermal Neutron Cross Section 0.430 barns/atom 0.430 barns/atom X-ray Absorption Edge 3.07016 Å 3.07016 Å K ...

### Physical and Chemical Properties of Calcium? | Yahoo Answers

Best Answer: I wish chemistry teachers would stop trying to throw this into the first few weeks of chemistry. In many cases the differences between a chemical change and a physical change are so blurry as to be nonexistent. Students simply don't have the experience and knowledge to know how to deal with this.

### Physical or Chemical Properties? | Yahoo Answers

Yahoo India Answers ... What is physical property and chemical property? 1 following . 5 answers 5. ... Answers. Relevance. Rating Newest Oldest. Best Answer: Physical properties: Properties that do not change the chemical nature of matter Chemical properties: Properties that do change the chemical nature of matter ...

### what is physical property and chemical property? | Yahoo ...

1.mass is the amount of matter in an object 2.the same volume 3.if your talking about a triple beam balance then it measures mass 4.graduated cylinder 5.Density 6.flexibility 7.Proton 8.

### Physical characteristics/properties ... - sg.answers.yahoo.com

When you look at oranges, you know that they are oranges because of their color, shape, and smell. Mass, color, shape, volume, and density are some physical properties. The answers to the question about the present are physical properties. Density is an important physical property. Density is the mass of a substance per unit volume.

### What are the properties of matter? | Yahoo Answers

Physical changes are changes in the condition of the material or changes caused by mixing materials together or taking them apart--just mixing or separating, not the creation of something new.If you take some water and you heat it up and it changes to gas, that is a physical change. You

still have water, only now it is steam instead of liquid.

**Physical or Chemical? | Yahoo Answers**

Yahoo New Zealand Answers ... Chemical properties are usually observed in the course of a chemical reaction, while physical properties are observed by examining a sample of the pure element. The chemical properties of an element are due to the distribution of electrons around the atom's nucleus, particularly the outer, or valence, electrons; it ...

**Physical and Chemical Properties of an element? | Yahoo ...**

Best Answer: Physical: Uranium is a silvery, shiny metal that is both ductile and malleable.  
Chemical: Uranium is a relatively reactive element. It combines with nonmetals such as oxygen, sulfur, chlorine, fluorine, phosphorus, and bromine. It also dissolves in acids and reacts with water.

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